CERTIFICATE OF ANALYSIS

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Issued by EffecTech Date of issue 24 July 2024

Certificate number 24/0206/11

A. Parc

Name: Alisha Patel

Signature



line temperature 11.4 °C, sampled by R. Bell, sample vessel #CB4044.

Date of analysis : 19 July 2024

This laboratory was not responsible for the sampling stage as the sample and sample identification information was provided by the customer. The results presented in this certificate apply to the sample analysed on an as received basis. **Composition**

component	amount fraction	component	amount fraction
	(%mol/mol)		(%mol/mol)
helium	$0.00743~\pm~0.00029$	benzene	$0.00039~\pm~0.00004$
hydrogen	$0.00184~\pm~0.00009$	cyclohexane	$0.00036~\pm~0.00004$
argon	$0.00432~\pm~0.00019$	heptanes†	$0.00055~\pm~0.00005$
oxygen	<0.0015	n-heptane	$0.00006~\pm~0.00002$
nitrogen	$0.8153~\pm~0.0036$	toluene	$0.00006~\pm~0.00002$
carbon dioxide	$1.9347~\pm~0.0047$	methylcyclohexane	$0.00011~\pm~0.00002$
methane	$90.268 ~\pm~ 0.022$	octanes†	$0.00002~\pm~0.00001$
ethane	$5.533~\pm~0.013$	n-octane	<0.00001
propane	1.1963 \pm 0.0042	nonanes†	‡<0.00001
iso-butane	$0.08841~\pm~0.00080$	n-nonane	<0.00001
n-butane	$0.12217 ~\pm~ 0.00080$	decanes†	‡<0.00001
neo-pentane	$0.00035~\pm~0.00004$	n-decane	<0.00001
iso-pentane	$0.01301~\pm~0.00026$	undecanes†*	‡<0.00001
n-pentane	$0.01072~\pm~0.00023$	n-undecane*	<0.00001
2-methylpentane	$0.00096~\pm~0.00007$	dodecanes†*	‡<0.00001
3-methylpentane	$0.00052~\pm~0.00005$	n-dodecane*	<0.00001
2,2-dimethylbutane	$0.00013 ~\pm~ 0.00002$		
hexanes†	$0.00060~\pm~0.00005$	C_6 + (hexanes+)§	$0.00465~\pm~0.00015$
n-hexane	$0.00089~\pm~0.00006$		

The reported expanded uncertainty is based on a standard uncertainty multiplied by a coverage factor k=2, which for a normal distribution provides a level of confidence of approximately 95%. The uncertainty evaluation has been carried out in accordance with JCGM 100:2008 - *Evaluation of measurement data* - *Guide to the expression of uncertainty in measurement* (GUM).

* these components/quantities are not UKAS accredited as they lie outside the scope of accreditation for our laboratory

† the amount fraction of a grouped component is the sum of all isomers in that group except for those identified separately.

‡ no individual isomers could be measured above this limit of detection.

§ the C₆₊ (hexanes+) component is the sum of amount fractions of all hydrocarbons containing 6 carbon atoms or greater.

EffecTech is accredited by UKAS to undertake the analysis presented in this certificate according to ISO/IEC 17025:2017. The laboratory accreditation requirements of the United Kingdom Accreditation Service. It provides traceability of measurement to the SI system of units and/or to units of measurement realised at the National Physical Laboratory or other is using laboratory. The laboratory accreditation requirements of the United Kingdom Accreditation Service. It provides traceability of measurement to the SI system of units and/or to units of measurement realised at the National Physical Laboratory or other recognised national metrology institutes. This certificate may not be reproduced other than in full, except with the prior written approval of the issuing laboratory. The laboratory activities reported were performed at the location of the issuing body The reference values reported relate only to the specific sample identified in this certificate

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UKAS accredited testing laboratory no.1927

Physical Properties

Reference conditions	primary combustion 15°C metering 15°C		secondary combustion 0 °C metering 0 °C	
mean molar mass compression factor	$\begin{array}{c} 17.899 \pm 0.018 \\ 0.9976 \pm 0.0010 \end{array}$	kg∙kmol⁻¹	$\begin{array}{c} 17.899 \pm 0.018 \\ 0.9971 \pm 0.0010 \end{array}$	kg∙kmol⁻¹
Real gas properties				
superior calorific value	$\begin{array}{c} 39.211 \pm 0.039 \\ 924.90 \pm 0.92 \\ 51.674 \pm 0.052 \end{array}$	MJ·m ⁻³ kJ·mol ^{−1} MJ·kg ⁻¹	$\begin{array}{c} 41.448 \pm 0.041 \\ 926.34 \pm 0.93 \\ 51.755 \pm 0.052 \end{array}$	MJ⋅m ⁻³ kJ⋅mol ⁻¹ MJ⋅kg ⁻¹
inferior calorific value	$\begin{array}{c} 35.384 \pm 0.035 \\ 834.63 \pm 0.83 \\ 46.631 \pm 0.047 \end{array}$	MJ·m ⁻³ kJ·mol⁻¹ MJ·kg⁻¹	$\begin{array}{c} 37.351 \pm 0.037 \\ 834.77 \pm 0.83 \\ 46.639 \pm 0.047 \end{array}$	MJ⋅m ⁻³ kJ⋅mol ⁻¹ MJ⋅kg ⁻¹
relative density density superior Wobbe index	$\begin{array}{c} 0.61923 \pm 0.00062 \\ 0.75881 \pm 0.00076 \\ 49.829 \pm 0.050 \end{array}$	kg⋅m ⁻³ MJ⋅m ⁻³	$\begin{array}{c} 0.61942 \pm 0.00062 \\ 0.80086 \pm 0.00080 \\ 52.664 \pm 0.053 \end{array}$	kg⋅m ⁻³ MJ⋅m ⁻³
Ideal gas properties				
superior calorific value	$\begin{array}{c} 39.116 \pm 0.039 \\ 924.90 \pm 0.92 \\ 51.674 \pm 0.052 \end{array}$	MJ·m ⁻³ kJ·mol ⁻¹ MJ·kg ⁻¹	$\begin{array}{c} 41.329 \pm 0.041 \\ 926.34 \pm 0.93 \\ 51.755 \pm 0.052 \end{array}$	MJ⋅m ⁻³ kJ⋅mol ⁻¹ MJ⋅kg ⁻¹
inferior calorific value	$\begin{array}{c} 35.299 \pm 0.035 \\ 834.63 \pm 0.83 \\ 46.631 \pm 0.047 \end{array}$	MJ⋅m ⁻³ kJ⋅mol ⁻¹ MJ⋅kg ⁻¹	$\begin{array}{c} 37.243 \pm 0.037 \\ 834.77 \pm 0.83 \\ 46.639 \pm 0.047 \end{array}$	MJ⋅m ⁻³ kJ⋅mol ⁻¹ MJ⋅kg ⁻¹
relative density density superior Wobbe index	$\begin{array}{c} 0.61800 \pm 0.00062 \\ 0.75698 \pm 0.00076 \\ 49.758 \pm 0.050 \end{array}$	kg⋅m ⁻³ MJ⋅m ⁻³	$\begin{array}{c} 0.61800 \pm 0.00062 \\ 0.79855 \pm 0.00080 \\ 52.572 \pm 0.053 \end{array}$	kg⋅m ⁻³ MJ⋅m ⁻³

Auxiliary properties

The following additional properties are calculated from equivalent composition in accordance with the UK Statutory Instrument 1996 No.551 Gas Safety (Management) Regulations 1996 - Regulation 8 (Schedule 3) - Content and other Characteristics of Gas - Part III - Interpretation

sooting index	0.5153
incomplete combustion factor	-0.4279

The physical properties above are calculated from composition at a reference pressure of 1.01325 bar and at the combustion and metering temperatures stated in accordance with the international standard ISO 6976:1995 - *Natural Gas - Calculation of calorific value, density, relative density and Wobbe index from composition* (including amendment No.1 - May 1998).

Note 1: In accordance with the recommendations of the international standard, the gas mixture is assumed dry (free from moisture).

Note 2: For the purpose of these calculations grouped components are given the property of the corresponding normal alkane.

The reported expanded uncertainty is based on a standard uncertainty multiplied by a coverage factor k=2, which for a normal distribution provides a level of confidence of approximately 95%. The uncertainty evaluation has been carried out in accordance with JCGM 100:2008 - *Evaluation of measurement data* - *Guide to the expression of uncertainty in measurement* (GUM).

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